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Key indicators

Single-crystal X-ray study T = 293 K Mean σ (C–C) = 0.002 Å R factor = 0.024 wR factor = 0.077 Data-to-parameter ratio = 14.4

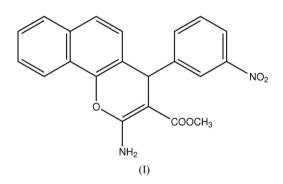
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

Methyl 2-amino-4-(3-nitrophenyl)-4*H*benzo[*h*]chromene-3-carboxylate

The title compound, $C_{21}H_{16}N_2O_5$, was synthesized by the reaction of 1-naphthol with methyl cyanoacetate and 3-nitrobenzaldehyde in methanol under microwave irradiation. Intermolecular C-H···O hydrogen bonds link the molecules into chains along the *b* axis and adjacent chains are connected *via* N-H···O hydrogen bonds.

Comment

Benzopyrans and their derivatives occupy an important place in the realm of natural and synthetic organic chemistry because of their biological and pharmacological properties (Morianka & Takahashi, 1977), such as antisterility (Brooks, 1998) and anticancer activities (Hyana & Saimoto, 1987). In addition, polyfunctionalized benzopyrans constitute the structural unit of a number of natural products and, because of the inherent reactivity of the inbuilt pyran ring, they may serve as versatile synthons (Hatakeyama *et al.*,1988). We report here the crystal structure of the title compound, (I).



In the molecule of (I) (Fig. 1), all bond lengths and angles are normal. An intramolecular $N-H\cdots O$ hydrogen bond (Table 1) determines the orientation of the carboxylate group. The pyran ring adopts a flattened envelope conformation, with atom C11 displaced by 0.151 (2) Å from the mean plane through atoms C1, C10, O1, C18 and C19. The dihedral angle between the C1-C10/O1/C18/C19 and C12-C17 planes is 82.30 (4)°.

In the crystal structure, intermolecular $C-H\cdots O$ hydrogen bonds link the molecules into chains along the *b* axis. Adjacent chains are connected *via* $N-H\cdots O$ hydrogen bonds (Table 1). A $C-H\cdots \pi$ interaction involving the C12-C17 benzene ring is also observed.

Experimental

© 2006 International Union of Crystallography All rights reserved Compound (I) was prepared by the reaction of 1-naphthol (5 mmol) with methyl cyanoacetate (5 mmol) and 3-nitrobenzaldehyde

Received 7 August 2006 Accepted 9 August 2006 (5 mmol) in methanol (2 ml) by using piperidine (0.5 mmol) as catalyst under microwave irradiation for 8 min. The pure compound (I) was obtained by recrystallization from methanol (m.p. 416–418 K). Crystals of (I) suitable for X-ray diffraction were obtained by slow evaporation of a methanol solution.

V = 932.8 (4) Å³

 $D_{\rm r} = 1.340 {\rm Mg} {\rm m}^{-3}$

Mo $K\alpha$ radiation

Block, colourless

 $0.40 \times 0.30 \times 0.20 \ \mathrm{mm}$

3 standard reflections

every 200 reflections

intensity decay: none

3658 independent reflections

2346 reflections with $I > 2\sigma(I)$

 $\mu = 0.10 \text{ mm}^-$ T = 293 (2) K

 $\begin{aligned} R_{\rm int} &= 0.063\\ \theta_{\rm max} &= 26.0^\circ \end{aligned}$

Z = 2

Crystal data

 $\begin{array}{l} C_{21}H_{16}N_2O_5 \\ M_r = 376.36 \\ \text{Triclinic, } P\overline{1} \\ a = 8.2190 \ (16) \ \mathring{A} \\ b = 9.2570 \ (19) \ \mathring{A} \\ c = 13.203 \ (3) \ \mathring{A} \\ \alpha = 71.84 \ (3)^\circ \\ \beta = 78.39 \ (3)^\circ \\ \gamma = 82.73 \ (3)^\circ \end{array}$

Data collection

Enraf-Nonius CAD-4 diffractometer $\omega/2\theta$ scans Absorption correction: ψ scan (North *et al.*, 1968) $T_{\min} = 0.963, T_{\max} = 0.984$ 3739 measured reflections

Refinement

Refinement on F^2	H-atom parameters constrained
$R[F^2 > 2\sigma(F^2)] = 0.024$	$w = 1/[\sigma^2(F_o^2) + (0.0249P)^2]$
$wR(F^2) = 0.077$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.04	$(\Delta/\sigma)_{\rm max} = 0.001$
3658 reflections	$\Delta \rho_{\rm max} = 0.25 \ {\rm e} \ {\rm \AA}^{-3}$
254 parameters	$\Delta \rho_{\rm min} = -0.11 \text{ e } \text{\AA}^{-3}$

Table 1	_	
Hydrogen-bond geometry	(Å,	°).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdot \cdot \cdot A$	$D - \mathbf{H} \cdots A$
$ \frac{1}{N1 - H1A \cdots O5^{i}} \\ N1 - H1B \cdots O2 \\ C15 - H15 \cdots O2^{ii} \\ C21 - H21A \cdots O5^{iii} \\ C21 - H21C \cdots Cg1^{iv} $	0.86	2.22	3.0750 (18)	173
	0.86	2.05	2.6616 (16)	127
	0.93	2.48	3.373 (2)	161
	0.96	2.57	3.504 (2)	165
	0.96	2.83	3.5675 (19)	134

Symmetry codes: (i) x - 1, y + 1, z; (ii) x, y - 1, z; (iii) x, y + 1, z; (iv) -x + 1, -y + 1, -z + 1. *Cg*1) is the centroid of the Cl2–Cl7 benzene ring.

All H atoms were placed in idealized positions and refined as riding, with C-H = 0.93–0.98 Å, N-H = 0.86 Å and $U_{iso}(H) = 1.2$ or 1.5 times U_{eq} (carrier atom).

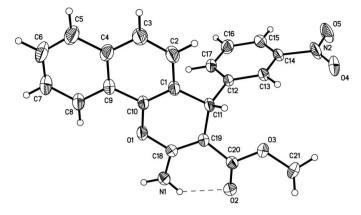


Figure 1

The molecular structure of (I), showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 30% probability level and the intramolecular hydrogen bond is indicated by a dashed line.

Data collection: *CAD-4 Software* (Enraf–Nonius, 1989); cell refinement: *CAD-4 Software*; data reduction: *XCAD4* (Harms & Wocadlo, 1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Siemens, 1996); software used to prepare material for publication: *SHELXTL*.

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References

- Brooks, G. T. (1998). Pestic. Sci. 22, 41-50.
- Enraf-Nonius (1989). *CAD-4 Software*. Version 5.0. Enraf-Nonius, Delft, The Netherlands.
- Harms, K. & Wocadlo, S. (1995). XCAD4. University of Marburg, Germany.
- Hatakeyama, S., Ochi, N., Numata, H. & Takano, S. (1988). J. Chem. Soc. Chem. Commun. pp. 1202–1024.
- Hyana, T. & Saimoto, H. (1987). Jpn Patent No. JP62l812768.
- Morianka, Y. & Takahashi, K. (1977). Jpn Patent No. JP52109000.
- North, A. C. T., Phillips, D. C. & Mathews, F. S. (1968). Acta Cryst. A24, 351–359.
- Sheldrick, G. M. (1997). SHELXL97 and SHELXS97. University of Göttingen, Germany.
- Siemens (1996). *SHELXTL*. Version 5.06. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.